



Sonopant Dandekar Shikshan Mandali's
Sonopant Dandekar Arts, V.S. Apte Commerce
& M.H. Mehta Science College, Palghar

VALUE ADDED COURSE
IN
BASIC CHEMISTRY

(F.Y.B.Sc.)

Report 2022-2023

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Value Added Course

Duration: 32 hours

Aim: To study basic concepts in organic, inorganic and physical chemistry

Features of the course:

- Basic concepts of Physical, Inorganic and Organic chemistry
- Practice of important concepts.
- Delivered by faculty of department of chemistry.

Objectives:

- To acquaint learner with fundamental concept of physical chemistry.
- By the end of the course students will gain knowledge about basics of thermodynamics, kinetics, electrochemistry, nuclear chemistry, solid state chemistry.
- Understand the principles and rules of IUPAC nomenclature for organic compounds.
- Familiarize yourself with the various types of reagents used in organic synthesis and their applications.
- Comprehend the electronic effects and their influence on organic reactions and molecular properties.
- Gain knowledge of stereochemical principles and their significance in organic chemistry.
- Understand basic principles in inorganic chemistry.
- Develop understanding of periodic table and arrangement of elements.

Eligibility:

Any science students having chemistry as a subject.

Proposal

20th April, 2023

To,
The Principal,
Sonopant Dandekar Arts, V.S. Apte Commerce
and M.H. Mehta Science College,
Palghar

Subject: Approval for Value Added Course in Basic Chemistry (F.Y.B.Sc.)

Respected Sir,

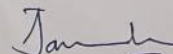
For the strong foundation in S.Y.B.Sc, basic concepts in chemistry needs to be revised and cleared for students. We have designed a value-added course in basics of chemistry for the student's opting chemistry in their second year. The course is aimed at enhancing knowledge and revising some basic concepts of chemistry. This course will benefit the students opting chemistry in S.Y.B.Sc. The details are provided along with this letter.

I request you to kindly grant the permission to start this course and look forward to your positive response.

Thanking you,

Sandona
Kumar
20/04/2023

Yours faithfully,



Dr. Suhas P. Janwadkar

Head, Department of Chemistry

Notice for students



Sonopant Dandekar Arts, V. S. Apte Commerce &
M. H. Mehta Science College, Palghar, Dist: Palghar
ACADEMIC YEAR 2022-23

Date : 02/05/2023

NOTICE

All the **F. Y. B. Sc.** students are hereby informed that their value-added course in basics of chemistry will begin from 16th May, 2023. Students opting chemistry in S. Y. B. Sc. are requested to join the following group,
<https://chat.whatsapp.com/IEQV5d3nDxX2Y8969LdW15>

The schedule of course is attached with this notice.

Dr. Suhas Janwadkar

Head ,Department of Chemistry

Syllabus

Syllabus: Inorganic Chemistry

Sr. No.	Topic	Content	Duration
1	Inorganic Chemistry	Introduction to Inorganic Chemistry	➤ 4 Hrs.
2	Electronic Configuration	Quantum Numbers	➤ 12 Hrs.
		Electronic Configuration, Aufbau Principle	
		Electronic Configuration	
		Pauli Exclusion Principle	
		Hund's Rule	
3	Periodic Table	Introduction to Periodic Table	➤ 10 Hrs.
		Periodic Table	
		Classification	
		Periodicity	
		Periodicity	
4	Valency and Oxidation State	Valency and Oxidation State	➤ 4 Hrs.
	Total Duration		➤ 30 Hrs.

Syllabus: Organic Chemistry

	Content	Theoretical	Duration
1	➤ IUPAC nomenclature	➤ IUPAC of various functional group	➤ 10 Hrs.
2	➤ Basic reaction mechanism	➤ Nucleophile, Electrophile and Types of arrows ➤ Various effects ➤ Types of reactions ➤ Intermediates ➤ Hybridization of C, O, N	➤ 01 Hrs. ➤ 04 Hrs. ➤ 01 Hrs. ➤ 03 Hrs. ➤ 01 Hrs.
3	➤ Stereochemistry	➤ Conformational isomers ➤ Configurational isomers ➤ Optical isomers	➤ 02 Hrs. ➤ 02 Hrs. ➤ 02 Hrs.
4	➤ Internal test	➤	➤ 04 Hrs
	Total Duration		➤ 30 Hrs.

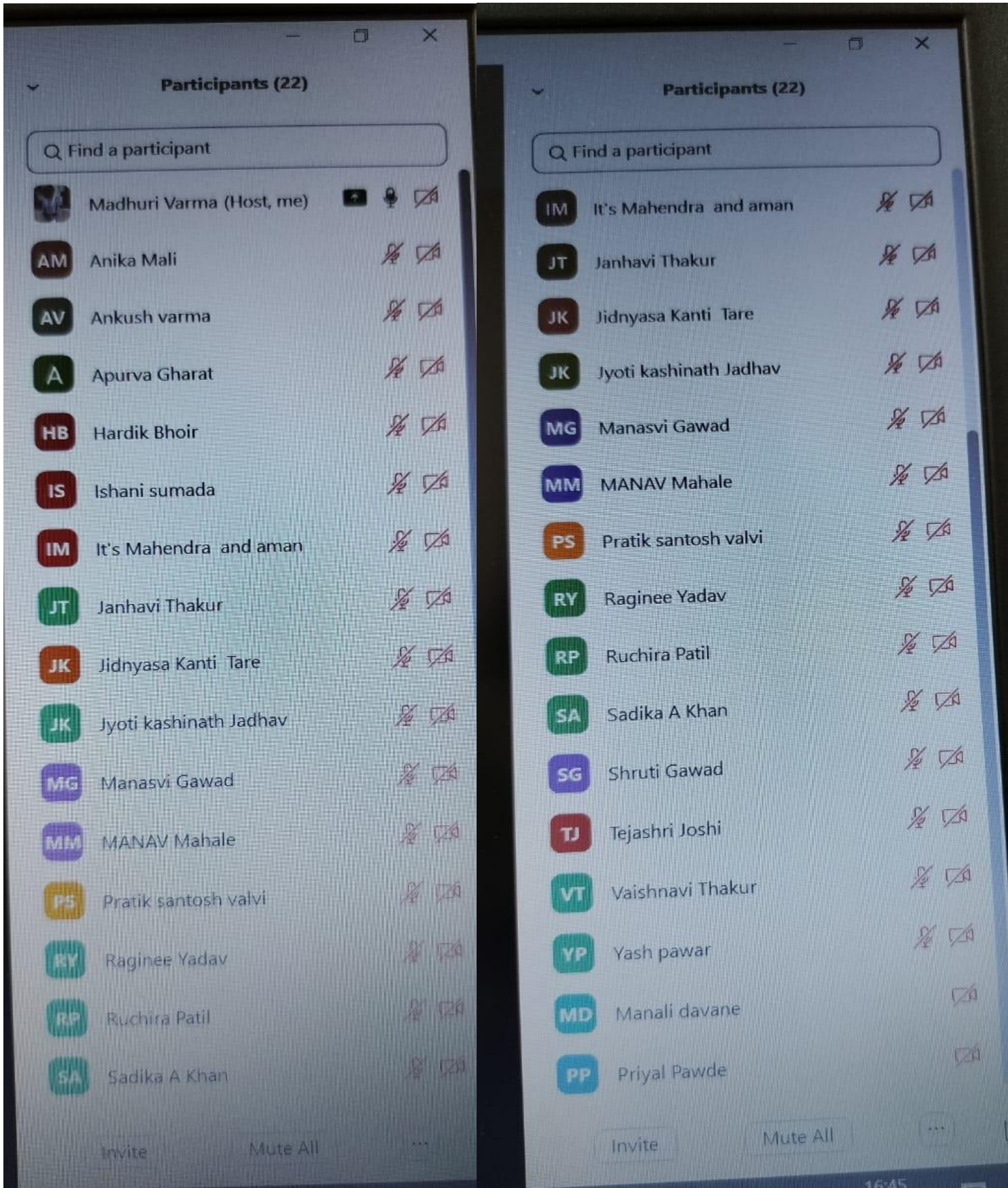
Syllabus

Syllabus: Physical Chemistry

	Content	Theoretical	Durations
1	➤ Chemical calculations	<ul style="list-style-type: none">➤ Mole Concept➤ Normality➤ Molarity➤ Formality➤ Molality➤ Mole Fraction➤ PPM➤ PPB	➤ 4 hours
2	➤ Chemical Kinetics	<ul style="list-style-type: none">➤ Rate of Reaction➤ Molecularity➤ Order of reaction➤ First Order reactions➤ Second order reaction (Derivations)	➤ 4 hours
3	➤ Chemical Thermodynamics	<ul style="list-style-type: none">➤ Extensive and intensive Properties➤ Basic terms involved in thermodynamics➤ First and second law of thermodynamics➤ Enthalpy➤ Entropy➤ Gibbs free energy	➤ 5 hours
4	➤ Electrochemistry	<ul style="list-style-type: none">➤ Electrolytic and Galvanic cells➤ Electrodes and electrolytes➤ Faraday's law➤ Kohlrauch Law➤ Type of cells➤ Use of salt bridge	➤ 5 hours
5	➤ Solutions	<ul style="list-style-type: none">➤ Types of solutions➤ Raoult's law	➤ 4 hours

		<ul style="list-style-type: none"> ➤ Vapour pressure of Solutions ➤ Deviations from Raoult's law ➤ Colligative properties 	
6	<ul style="list-style-type: none"> ➤ Nuclear Chemistry 	<ul style="list-style-type: none"> ➤ Nuclear Radioactivity ➤ Nuclear Transmutation reaction ➤ Nuclear fission ➤ Nuclear fusion 	<ul style="list-style-type: none"> ➤ 4 hours
7	<ul style="list-style-type: none"> ➤ Solid State Chemistry 	<ul style="list-style-type: none"> ➤ Law's of Crystallography ➤ Crystal Lattice ➤ Body centered Cubic lattice ➤ Face centered Cubic lattice ➤ Miller indices ➤ Numericals 	<ul style="list-style-type: none"> ➤ 4 hours

Attendance



[Close](#) **Participants (22)**

- MV** Madhuri Varma (me)
- Madhuri Varma (Host)
- AM** Anika Mali
- AV** Ankush varma
- HB** Hardik Bhoir
- HK** Husna khan
- IS** Ishani sumada
- JT** Janhavi Thakur
- JK** Jidnyasa Kanti Tare
- MD** Manali davane
- NV** Neetish Vrma

[Close](#) **Participants (22)**

- NV** Neetish Vrma
- PT** Pranchita Tare
- P** Pratik valvi
- PP** Priyal Pawde
- R** Ravita
- RP** Ruchira Patil
- S9** Sahil 90021
- SM** Shritija More
- SG** Shruti Gawad
- SG** Sneha Gowari
- VT** Vaishnavi Thakur
- YP** Yash pawar

[Invite](#)

[Invite](#)

Glimpse of event

Mendeleev's Periodic Table (1869)

I	II	III	IV	V	VI	VII	VIII		
H 1.01	Be 9.01	B 10.8	C 12.0	N 14.0	O 16.0	F 19.0			
Li 6.94	Mg 24.33	Al 27.0	Si 28.1	P 31.0	S 32.1	Cl 35.5			
K 39.1	Ca 40.1		Ti 47.9	V 50.9	Cr 52.0	Mn 54.9	Fe 55.9	Co 58.9	Ni 58.7
Cu 63.5	Zn 65.4			As 74.9	Se 79.0	Br 79.9			
Rb 85.5	Sr 87.6	Y 88.9	Zr 91.2	Nb 92.9	Mo 95.9		Ru 101	Rh 103	Pd 106
Ag 108	Cd 112	In 115	Sn 119	Sb 122	Te 128	I 127			
Ce 133	Ba 137	La 139		Ta 181	W 184				
Au 197	Hg 201	Tl 204	Pb 207	Bi 209					

Palghar, Maharashtra, India
 PQ23+7RW, Ganesh Nagar, Tembhode, Palghar,
 Maharashtra 401404, India
 Lat 19.700888°
 Long 72.754595°
 31/05/23 11:50 AM GMT +05:30

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- $M = \frac{\text{Number of mill moles of solute}}{\text{volume of solution in mL}}$
- $M = \frac{10 \times \text{density} \times \text{weight percentage}}{\text{molecular weight of solute}}$
- $M = \frac{10 \times d \times \%(\text{w/w})}{M_2}$

Participants (21)

- Madhuri Varma (Host, me)
- AM Anika Mali
- AV Ankush varma
- A Apurva Gharat
- HB Hardik Bhoir
- IS Ishani sumada
- IM It's Mahendra. and aman
- JT Janhavi Thakur
- JK Jidryasa Kanti Tare
- JK Jyoti kashinath Jadhav
- MM MANAV Mahale
- PS Pratik santosh valvi
- PP Priyal Pawde
- RV Raginee Yadav
- RP Ruchira Patil
- SM Shreelja More

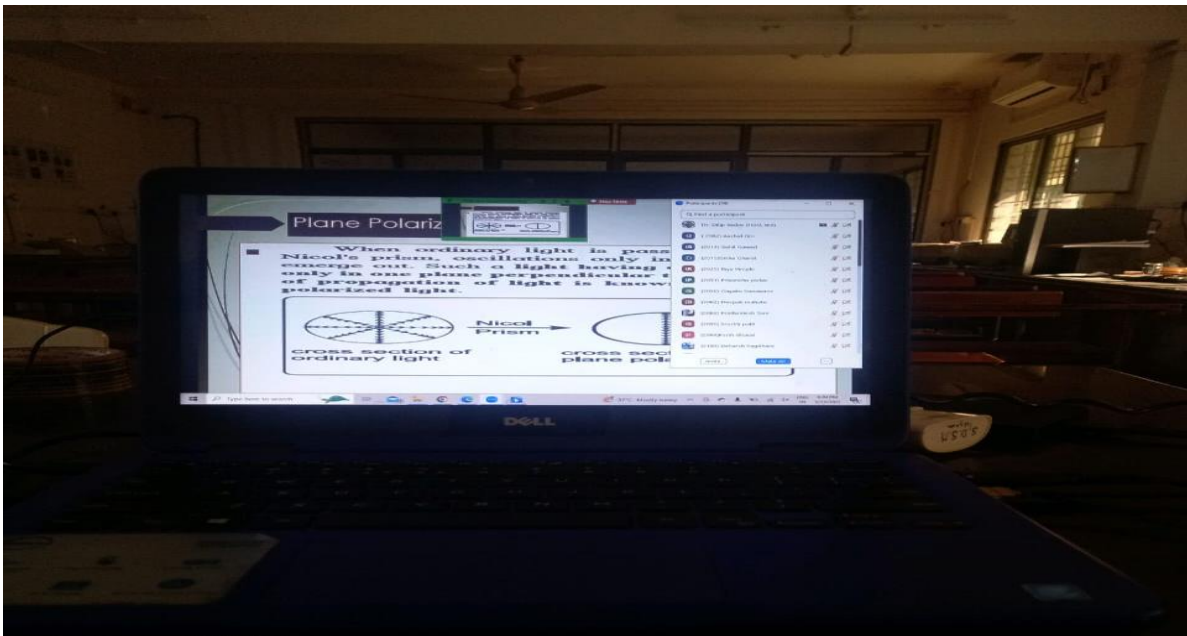


Palghar Bhaji Market, Juna Palghar, Palghar, Maharashtra 401404, India

Palghar
Maharashtra
India

2023-05-24(Wed) 04:45(pm)

31°C
88°F

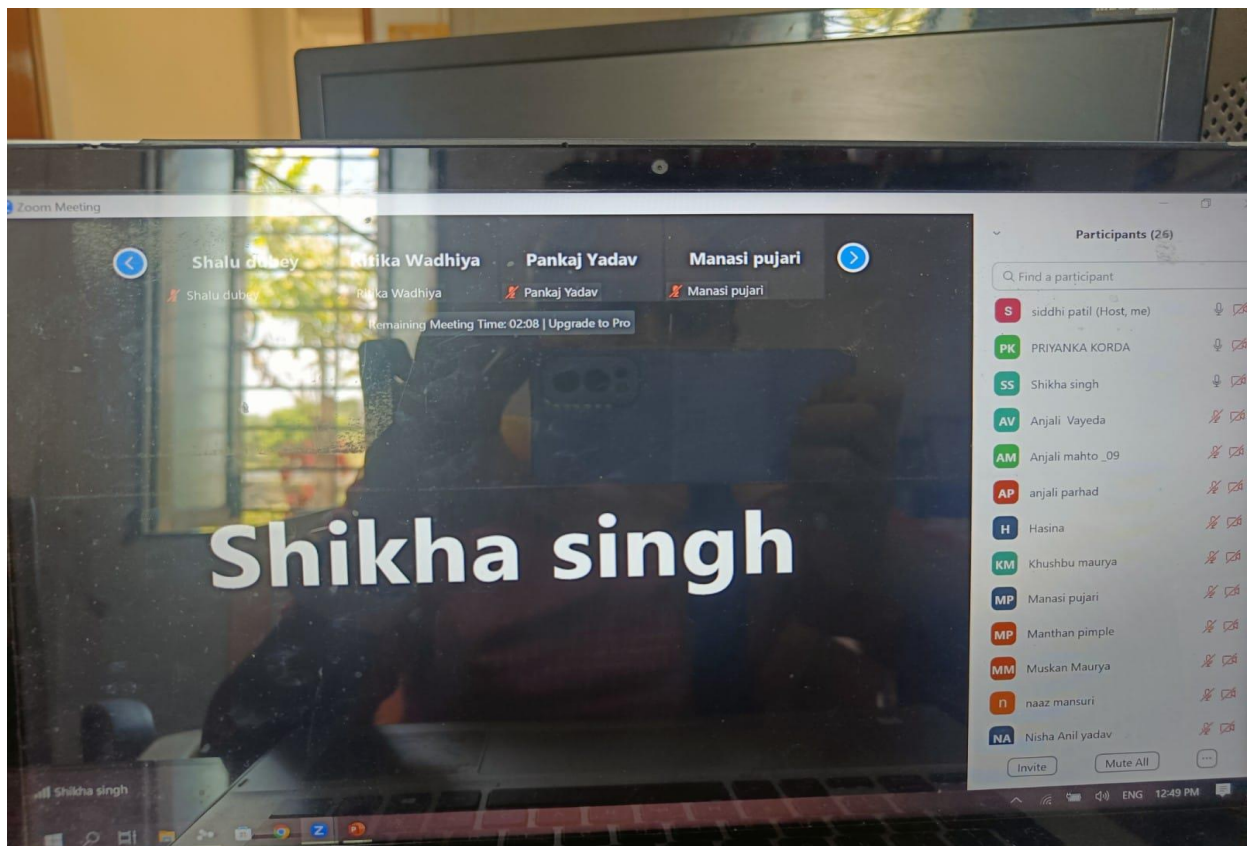
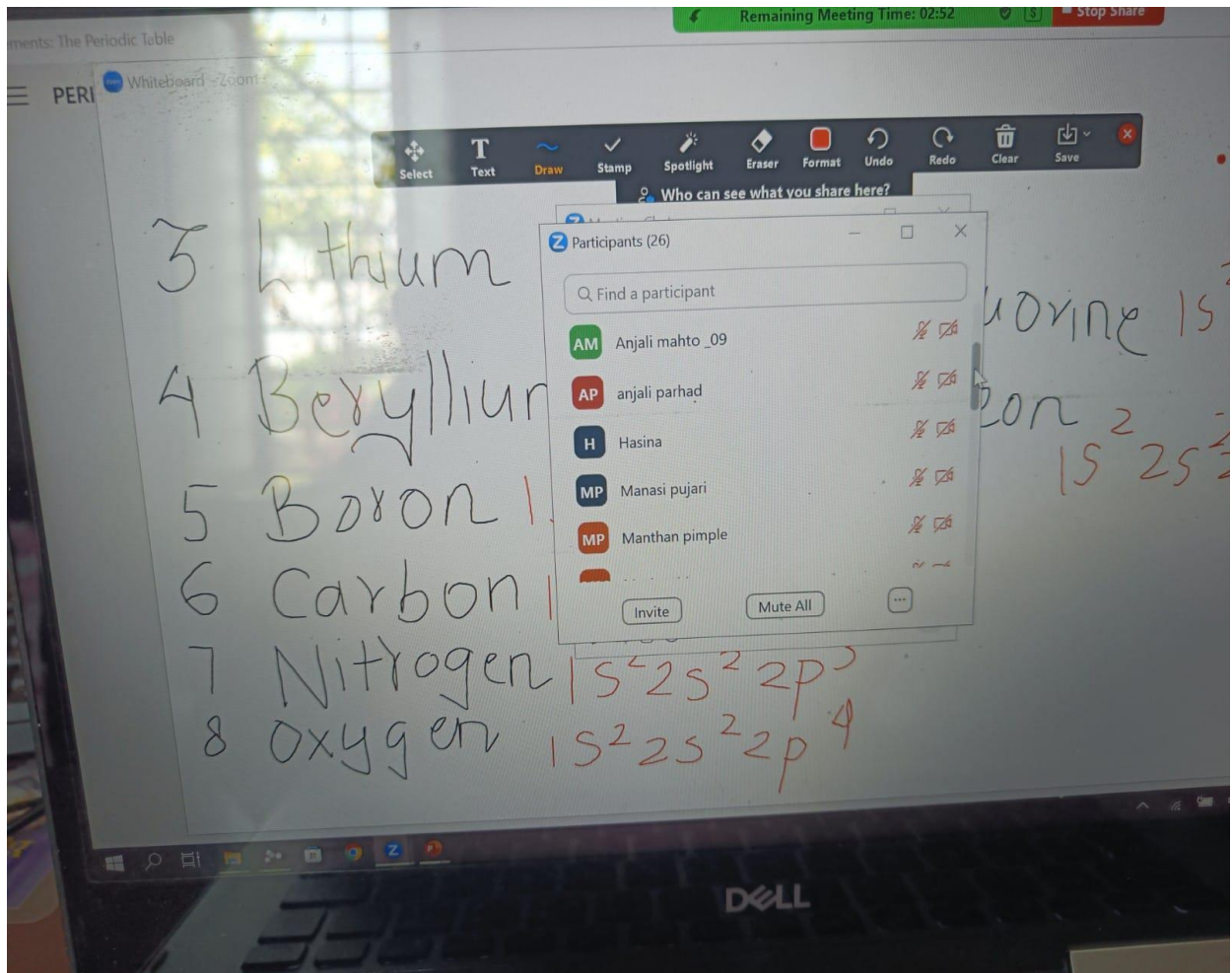


Palghar Bhaji Market, Juna Palghar, Palghar, Maharashtra 401404, India

Palghar
Maharashtra
India

2023-05-23(Tue) 04:19(pm)

32°C
90°F



Course Summary

The Value Added Course in Chemistry, offered over a span of 32 hours, aimed to provide students with a comprehensive understanding of fundamental concepts across physical, inorganic, and organic chemistry. Delivered by the experienced faculty of the Department of Chemistry, the course was structured to blend theoretical knowledge with practical insights, ensuring a holistic learning experience.